

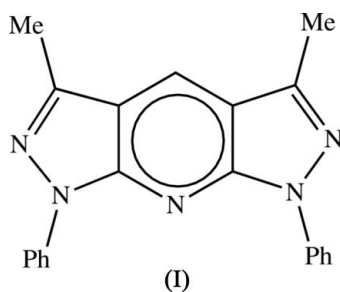
Jaime Portilla,<sup>a</sup> Jose M. de la Torre,<sup>b</sup> Justo Cobo,<sup>b</sup> John N. Low<sup>c</sup> and Christopher Glidewell<sup>d\*</sup><sup>a</sup>Grupo de Investigación de Compuestos Heterocíclicos, Departamento de Química, Universidad de Valle, A.A. 25360 Cali, Colombia, <sup>b</sup>Departamento de Química Inorgánica y Orgánica, Universidad de Jaén, 23071 - Jaén, Spain, <sup>c</sup>Department of Chemistry, University of Aberdeen, Meston Walk, Old Aberdeen, AB24 3UE, Scotland, and <sup>d</sup>School of Chemistry, University of St Andrews, St Andrews, Fife, KY16 9ST, Scotland

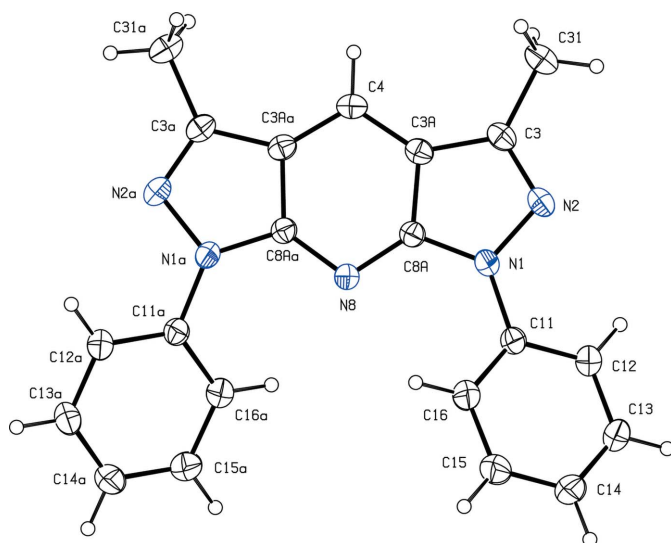
Correspondence e-mail: cg@st-andrews.ac.uk

## Key indicators

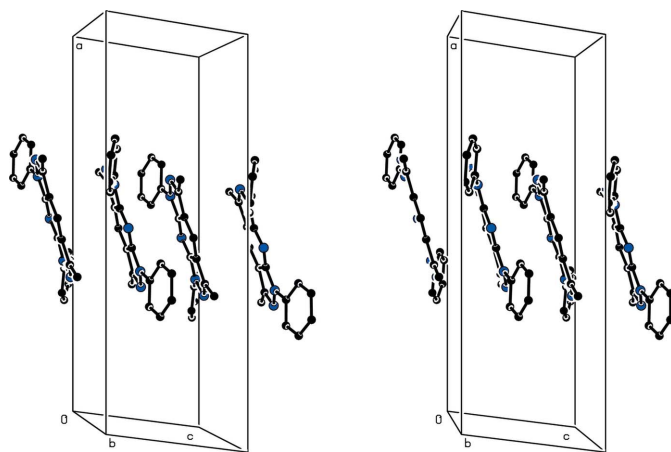
Single-crystal X-ray study  
 $T = 293$  K  
Mean  $\sigma(\text{C}-\text{C}) = 0.002$  Å  
 $R$  factor = 0.042  
 $wR$  factor = 0.116  
Data-to-parameter ratio = 16.3For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>. $\pi$ -Stacked chains in 3,5-dimethyl-1,7-diphenyl-1,7-dihydrodipyrzolo[3,4-*b*,4',3'-*e*]pyridineThe title compound,  $\text{C}_{21}\text{H}_{17}\text{N}_5$ , was prepared using a microwave-induced condensation reaction between 5-amino-3-methyl-1-phenylpyrazole and formaldehyde. The molecules lie across twofold rotation axes in space group  $C2/c$  and are into chains by a  $\pi$ - $\pi$  stacking interaction.Received 3 March 2006  
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## Comment

The title compound, (I), has been prepared using microwave irradiation in a solvent-free system and this provides an attractive alternative to the method recently reported (Abramos *et al.*, 2001), not only in eliminating the solvent, but also in reducing the reaction time from hours to minutes while considerably improving the yield, from 37% to 65%. The simplicity of the present procedure and its selectivity also contrast with the previous method which required two distinct azoles, an aminopyrazole and 5-chloro-4-formylpyrazole, to generate the product.The molecules of the title compound (I) (Fig. 1) lie across twofold rotation axes in space group  $C2/c$ : the reference molecule was selected as that lying across the axis along  $(\frac{1}{2}, y, \frac{1}{4})$ .The bond distances (Table 1) within the pyridine ring are consistent with aromatic delocalization, but there is very strong bond fixation within the pyrazole rings (see scheme). The dihedral angle between the phenyl ring and the adjacent pyrazole ring is  $27.4(2)^\circ$ .A single  $\pi \cdots \pi$  stacking interaction links the molecules into chains. The reference molecule, which lies across  $(\frac{1}{2}, y, \frac{1}{4})$ , is related by inversion to the adjacent molecules lying across the axes along  $(\frac{1}{2}, y, -\frac{1}{4})$  and  $(\frac{1}{2}, y, \frac{3}{4})$ ; the heterocyclic systems in these three molecules are thus parallel with an interplanar spacing between adjacent rings of  $3.363(2)$  Å. The ring centroid separations between the pyridine ring of the reference molecule and the pyridine and pyrazole rings of an adjacent molecule are  $3.772(2)$  Å and  $3.489(2)$  Å, respectively. Propagation of this interaction by inversion thus generates a chain of  $\pi$ -stacked molecules along the [001]



**Figure 1**  
The molecular structure of compound (I), showing the atom-labelling scheme. Displacement ellipsoids are drawn at the 30% probability level, and the atoms marked 'a' are at the symmetry position ( $1 - x, y, \frac{1}{2} - z$ ).



**Figure 2**  
A stereoview of part of the crystal structure of (I), showing the formation of a  $\pi$ -stacked chain along [001]. For the sake of clarity, H atoms have been omitted.

direction (Fig. 2). Two chains of this type, related to one another by the  $C$ -centring operation, pass through each unit cell, but there are no direction-specific interactions between adjacent chains: in particular  $C-H \cdots N$  and  $C-H \cdots \pi$  hydrogen bonds are absent from the structure of (I).

## Experimental

Equimolar amounts of 5-amino-3-methyl-1-phenylpyrazole (1.0 mmol) and formaldehyde (1.0 mmol as 37% aqueous solution) were placed in open Pyrex vessels and irradiated in a domestic microwave oven for 1.5 min at 600 W. The reaction mixture was then extracted with ethanol. After the solvent had been removed under reduced pressure, the product was recrystallized from dimethyl-

formamide to give crystals that were suitable for single-crystal X-ray diffraction. Yield 65%; m. p. 485–486 K, literature value 490–491 K (Abramos *et al.*, 2001),

### Crystal data

$C_{21}H_{17}N_5$   
 $M_r = 339.40$   
Monoclinic,  $C2/c$   
 $a = 21.2976$  (4) Å  
 $b = 10.9267$  (3) Å  
 $c = 7.4201$  (2) Å  
 $\beta = 98.108$  (2)°  
 $V = 1709.49$  (7) Å<sup>3</sup>  
 $Z = 4$

$D_x = 1.319$  Mg m<sup>-3</sup>  
Mo  $K\alpha$  radiation  
Cell parameters from 1957 reflections  
 $\theta = 3.4$ – $27.5$ °  
 $\mu = 0.08$  mm<sup>-1</sup>  
 $T = 293$  (2) K  
Lath, colourless  
 $0.60 \times 0.18 \times 0.12$  mm

### Data collection

Bruker–Nonius KappaCCD diffractometer  
 $\varphi$  and  $\omega$  scans  
Absorption correction: multi-scan (SADABS; Sheldrick, 2003)  
 $T_{\min} = 0.925$ ,  $T_{\max} = 0.990$   
14999 measured reflections

1957 independent reflections  
1703 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.027$   
 $\theta_{\text{max}} = 27.5$ °  
 $h = -27 \rightarrow 27$   
 $k = -14 \rightarrow 13$   
 $l = -8 \rightarrow 9$

### Refinement

Refinement on  $F^2$   
 $R[F^2 > 2\sigma(F^2)] = 0.042$   
 $wR(F^2) = 0.116$   
 $S = 1.05$   
1957 reflections  
120 parameters  
H-atom parameters constrained

$w = 1/[\sigma^2(F_o^2) + (0.0625P)^2 + 0.6017P]$   
where  $P = (F_o^2 + 2F_c^2)/3$   
 $(\Delta/\sigma)_{\text{max}} < 0.001$   
 $\Delta\rho_{\text{max}} = 0.21$  e Å<sup>-3</sup>  
 $\Delta\rho_{\text{min}} = -0.17$  e Å<sup>-3</sup>

**Table 1**

Selected bond lengths (Å).

N1–N2	1.3900 (12)	N1–C8A	1.3686 (13)
N2–C3	1.3111 (15)	C3A–C8A	1.4195 (15)
C3–C3A	1.4320 (16)	N8–C8A	1.3375 (12)
C3A–C4	1.3860 (14)	N8–C8A	1.3375 (12)

All H atoms were located in difference maps, and then treated as riding atoms with  $C-H$  distances 0.93 Å (aromatic) or 0.96 Å (methyl), and with  $U_{\text{iso}}(H) = 1.2U_{\text{eq}}(C)$ , or  $1.5U_{\text{eq}}(C)$  for the methyl group.

Data collection: *COLLECT* (Hooft, 1999); cell refinement: *DENZO* (Otwinowski & Minor, 1997) and *COLLECT*; data reduction: *DENZO* and *COLLECT*; program(s) used to solve structure: *SIR2004* (Burla *et al.*, 2004) and *WinGX* (Farrugia, 1999); program(s) used to refine structure: *OSCAIL* (McArdle, 2003) and *SHELXL97* (Sheldrick, 1997); molecular graphics: *PLATON* (Spek, 2003); software used to prepare material for publication: *SHELXL97* and *PRPKAPPA* (Ferguson, 1999).

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